

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
3 May 2001 (03.05.2001)

PCT

(10) International Publication Number  
**WO 01/30308 A1**

(51) International Patent Classification<sup>7</sup>: A61K 7/00

Inc., 45 River Road, Edgewater, NJ 07020 (US). **NASER, Mark, Stephen**; Unilever Research U.S. Inc., 45 River Road, Edgewater, NJ 07020 (US). **CHAM, Philippe**; 204 Elder Avenue, Bergenfield, NJ 07621 (US).

(21) International Application Number: PCT/EP00/09762

(74) Agents: **ROTS, Maria, Johanna, Francisca et al.**; Unilever PLC. Patent Department, Colworth House, Sharnbrook, Bedford, Bedfordshire MK44 1LQ (GB).

(22) International Filing Date: 4 October 2000 (04.10.2000)

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW.

(25) Filing Language: English

(84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

(26) Publication Language: English

**Published:**

- With international search report.
- Before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments.

(30) Priority Data:  
09/425,699 22 October 1999 (22.10.1999) US

(71) Applicant (*for AE, AG, AU, BB, BZ, CA, CY, GB, GD, GH, GM, IE, IL, KE, LC, LK, LS, MN, MW, NZ, SD, SG, SI, SZ, TT, TZ, UG, ZA, ZW only*): **UNILEVER PLC [GB/GB]**; Unilever House, Blackfriars, London EC4P 4BQ (GB).

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(71) Applicant (*for all designated States except AE, AG, AU, BB, BZ, CA, CY, GB, GD, GH, GM, IE, IL, IN, KE, LC, LK, LS, MN, MW, NZ, SD, SG, SI, SZ, TT, TZ, UG, ZA, ZW*): **UNILEVER NV [NL/NL]**; Weena 455, NL-3013 AL Rotterdam (NL).

(71) Applicant (*for IN only*): **HINDUSTAN LEVER LIMITED [IN/IN]**; Hindustan Lever House, 165/166 Backbay Reclamation, Maharashtra, 400 020 Mumbai (IN).

(72) Inventors: **ST LEWIS, Dale**; Unilever Research U.S. Inc., 45 River Road, Edgewater, NJ 07020 (US). **KNAGGS, Helen, Elizabeth**; Unilever Research U.S.

WO 01/30308 A1

(54) Title: DUAL CHAMBER CLEANSING SYSTEM COMPRISING MULTIPLE EMULSION

(57) Abstract: The invention provides an at least dual compartment container containing at least one surfactant stripe and at least one multiple emulsion stripe. By dispensing a multiple emulsion stripe, greater deposition, particularly deposition of water soluble benefit agent (e.g., glycolic acid) is achieved than believed possible.

- 1 -

DUAL CHAMBER CLEANSING SYSTEM COMPRISING MULTIPLE EMULSION

The present invention relates to a composition for depositing water soluble benefit agents (e.g., glycolic acid, lactic acid, hydroxy caprylic acid, water soluble vitamins etc.) in greater amounts than previously possible when using personal wash shower gel type liquids and/or facial cleansers while maintaining good foam stability.

Specifically, the invention relates to a dual chamber system

10 in which one composition/stripe comprises a surfactant; and a separate composition/stripe, which is co-dispensed, comprises a multiple emulsion which in turn comprises a benefit agent emulsion (e.g., water soluble glycolic acid in oil) emulsified in a water solution containing additional

15 water soluble benefit agent (e.g., glycerol, PEGs, PPG, etc.)

It is greatly desirable to deposit water soluble benefit agents (e.g., glycolic acid; lactic acid) onto the skin or

20 another substrate.

However, deposition of water soluble benefit agent is extremely difficult to accomplish, particularly from wash-off type products (e.g., shower gels) because the water

25 soluble benefit agent will essentially wash off when the user rinses with water. In general, although there are still problems with shower gel products, it is much easier to deposit a hydrophobic benefit agent (e.g., silicone or petrolatum) than a hydrophilic one.

- 2 -

Unexpectedly, the applicants have found that, by forming a multiple emulsion of a water soluble benefit agent and separately dispensing the water soluble benefit agent containing multiple emulsion in one stripe and a surfactant 5 containing composition in another, the applicants have been able to deposit greater amounts of water soluble benefit agent than otherwise achievable (e.g., either through a single stripe cleanser, or through dual stripe cleanser having only oil in water emulsion, rather than the water-in-oil-in-water emulsion of invention).  
10

The use of separate surfactant and benefit agent stripes is not itself new. U.S. Patent No. 5,612,307 to Chambers et al., for example, teaches a dual chamber package comprising 15 separate surfactant and benefit agent stripe. The benefit agent, however, is a lipophilic benefit agent rather than a water-soluble one, and further is not in the multiple emulsion form of the multiple emulsion stripe of the invention.

20 Multiple emulsion like those of the invention are taught in the applicants' co-pending application entitled "Stable Multiple Emulsion Composition" to Naser et al. filed on the same date as the subject application, but this reference 25 does not teach the dual chamber dispensing system of the subject invention.

Unexpectedly, the applicants have found that a dual chamber dispenser comprising a surfactant containing stripe on one 30 side and a multiple emulsion containing stripe on the other (wherein the emulsion is a water in oil in water emulsion,

- 3 -

and wherein the internal water phase contains at least one water soluble benefit agent) is able to deliver water soluble benefit agent to the skin or other substrate in greater amounts than otherwise possible, i.e., using single  
5 stripe, or using different type of emulsion in the two stripe system such as oil in water only.

Specifically, the invention comprises:

- 10 - an aqueous liquid cleansing and moisturising composition comprising an at least dual chamber dispenser (in theory more than two stripes may be dispensed, and the invention is not necessarily limited by the number of stripes)  
comprising:  
15 (A) 10 to 99.9% by wt., preferably 30 to 99.9%, more preferably 50 to 99.9% of a surfactant containing stripe wherein about 1 to 75%, preferably 5 to 70% of said surfactant stripe comprises a surfactant selected from the group consisting of anionic, nonionic, zwitterionic and cationic surfactants, soap and mixtures thereof (water, solute, opacifier, bactericides and other standard ingredients may also be found in the stripe); and  
20 (B) 0.1 to 90%, preferably 0.1 to 70%, more preferably 0.1 to 50% by wt. of a multiple emulsion stripe comprising:  
25  
30

- 4 -

- (1) 1 to 99%, preferably 2 to 90% of said multiple emulsion of a water in oil ( $W_1/O$ ) emulsion comprising:
- 5 (a) about 1 to 99% of an internal aqueous phase comprising water, optional solute (0.01 to 30% solute) and optional surfactant (0 to 30%);
- 10 (b) 0.5 to 99%, preferably 1 to 80% of the emulsion of an oil phase surrounding said internal aqueous phase comprising a non-volatile silicone compound, a volatile hydrocarbon compound, a non-volatile hydrocarbon compound or a mixture thereof;
- 15 (c) about 0.1 to 20%, preferably 1 to 15% of a low HLB emulsifier (e.g., HLB under 10); and
- 20 (d) an effective amount (e.g., 0.01 to 40%, preferably 0.05 to 15%) of a topically effective water-soluble compound (e.g., glycolic acid) found in the internal aqueous phase; and
- (2) 1 to 99%, preferably 20 to 95% of said multiple emulsion of an external aqueous phase  $W_2$  comprising:
- 25 (a) 0 to 30%, preferably 0.01 to 5% of a high (10-30) HLB nonionic surfactant (e.g., Brij 58) which may be used to stop oil in the oil phase from leaking out of the  $W_2$  phase to form separate oil layer;

- 5 -

- (b) 0 to 60%, preferably 10 to 50% second topically active compound (e.g., glycerin);
- (c) optional solute; and
- 5 (d) 0 to 20% optional cleansing surfactant or surfactants wherein, if anionic, the surfactant is a non-amido surfactant.

As noted, the W<sub>2</sub> phase could be nothing but water.

10 Also, as noted in applicants co-pending application filed on same date, if predominantly anionic (>60%) in W<sub>2</sub> phase, a nonionic gum polymer is preferably present in W<sub>2</sub> and, if W<sub>2</sub> is predominantly amphoteric (>50%), an anionic gum polymer 15 is preferably present.

The present invention relates to an at least dual chamber dispenser comprising a surfactant containing stripe on one side ((A) above) and a multiple emulsion containing stripe 20 ((B) above) on the other. The multiple emulsion is a water in oil in water emulsion (W<sub>1</sub>-O-W<sub>2</sub>) wherein the internal water phase contains a water soluble benefit agent (e.g., glycolic acid). Each of the various components is described in greater detail below.

25 One stripe of the dispenser of the invention is the surfactant stripe (Component A). The surfactant containing stripe is really not limited in any way and any viable surfactant system may be used, although preferably this will 30 be a "mild" surfactant system.

- 6 -

The surface active agent can be selected from any known surfactant suitable for topical application to the human body. As noted, mild surfactants, i.e., surfactants which do not damage the stratum corneum, the outer layer of skin,  
5 are particularly preferred.

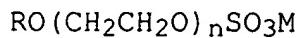
On preferred anionic surfactant is fatty acyl isethionate of formula:



where R is an alkyl or alkenyl group of 7 to 21 carbon atoms and M is a solubilizing cation such as sodium, potassium, ammonium or substituted ammonium. Preferably at least three  
15 quarters of the RCO groups have 12 to 18 carbon atoms and may be derived from coconut, palm or a coconut/palm blend.

Another preferred anionic surfactant is alkyl ether sulphate of formula:

20

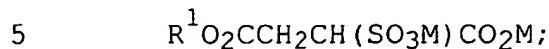


where R is an alkyl group of 8 to 22 carbon atoms, n ranges from 0.5 to 10 especially from 1.5 to 8, and M is a  
25 solubilizing cation as before.

Other possible anionic surfactants include alkyl glyceryl ether sulphate, sulphosuccinates, taurates, sarcosinates, sulphoacetates, alkyl phosphate, alkyl phosphate esters and  
30 acyl lactylate, alkyl glutamates and mixtures thereof.

- 7 -

Sulphosuccinates may be monoalkyl sulphosuccinates having the formula:

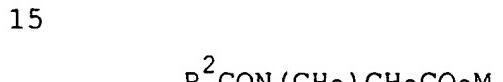


and amino-MEA sulphosuccinates of the formula:



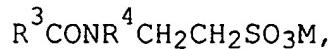
wherein  $R^1$  ranges for C<sub>8</sub>-C<sub>20</sub> alkyl, preferably C<sub>12</sub>-C<sub>15</sub> alkyl and M is a solubilizing cation.

Sarcosinates are generally indicated by the formula:



wherein  $R^2$  ranges from C<sub>8</sub>-C<sub>20</sub> alkyl, preferably C<sub>12</sub>-C<sub>15</sub> alkyl and M is a solubilizing cation.

20           Taurates are generally identified by the formula:



25           wherein  $R^3$  ranges from C<sub>8</sub>-C<sub>20</sub> alkyl preferably C<sub>12</sub>-C<sub>15</sub> alkyl,  $R^4$  ranges from C<sub>1</sub>-C<sub>4</sub> alkyl, and M is a solubilizing cation.

- 8 -

Harsh surfactants such as primary alkane sulphonate or alkyl benzene sulphonate will generally be avoided.

Suitable nonionic surfactants include alkyl polysaccharides,  
5 lactobionamides, ethylene glycol esters, glycerol monoethers, polyhydroxyamides (glucamide), primary and secondary alcohol ethoxylates, especially the C<sub>8</sub>-C<sub>20</sub> aliphatic alcohols ethoxylated with an average of from 1 to 20 moles of ethylene oxide per mole of alcohol.

10

If the surfactant comprises soap, the soap is preferably derived from materials with a C<sub>8</sub> to C<sub>22</sub> substantially saturated carbon chain and, preferably, is a potassium soap with a C<sub>12</sub> to C<sub>18</sub> carbon chain.

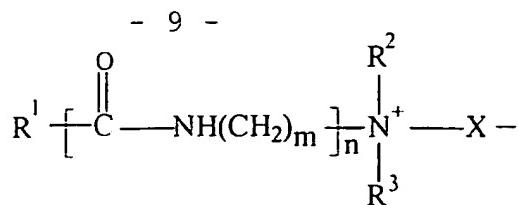
15

Mixtures of any of the foregoing surfactants may also be used.

The surfactant is preferably present at a level of from 1 to  
20 75 wt.%, preferably 5 to 70 wt.%.

It is also preferable that the composition includes from 0.5 to 15 wt.% of a co-surfactant agent with skin mildness benefits. Suitable materials are zwitterionic surfactants  
25 which have an alkyl or alkenyl group of 7 to 18 carbon atoms and comply with an overall structural formula:

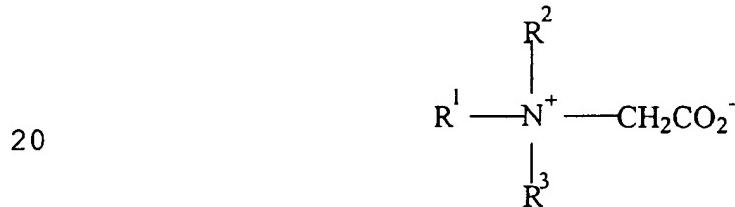
30



5 where  $\text{R}^1$  is alkyl or alkenyl of 7 to 18 carbon atoms  $\text{R}^2$  and  
 $\text{R}^3$  are each independently alkyl, hydroxyalkyl or carboxyalkyl of 1 to 3 carbon atoms;  
 $m$  is 2 to 4;  
 $n$  is 0 or 1;

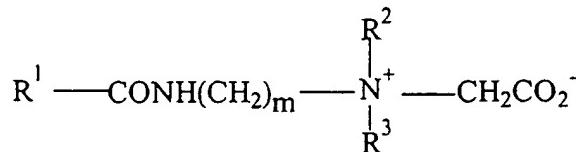
10  $\text{X}$  is alkylene of 1 to 3 carbon atoms optionally substituted with hydroxyl; and  
 $\text{Y}$  is  $-\text{CO}_2$  or  $-\text{SO}_3$ .

Zwitterionic surfactants within the above general formula  
15 include simple betaines of formula:



and amido betaines of formula:

25



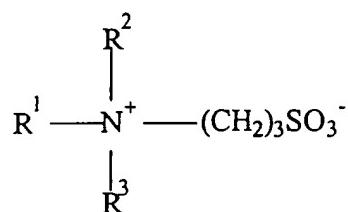
30 where  $m$  is 2 or 3.

- 10 -

In both formulae R<sup>1</sup>, R<sup>2</sup> and R<sup>3</sup> are as defined previously. R<sup>1</sup> may, in particular, be a mixture of C<sub>12</sub> and C<sub>14</sub> alkyl groups derived from coconut so that at least half, preferably at least three quarters, of the group R<sup>1</sup> has 10 to 14 carbon atoms. R<sup>2</sup> and R<sup>3</sup> are preferably methyl.

A further possibility is a sulphobetaine of formula:

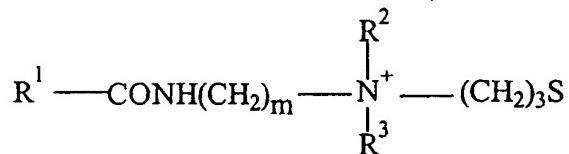
10



15

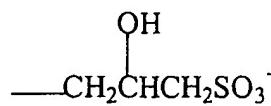
or

20



where m is 2 or 3, or variants of these in which -(CH<sub>2</sub>)<sub>3</sub>SO<sub>3</sub><sup>-</sup> is replaced by

25



- 11 -

R<sup>1</sup>, R<sup>2</sup> and R<sup>3</sup> in these formulae are as defined previously.

A structurant may be added to the phase comprising the surfactant. Suitable materials include swelling clays, for 5 example laponite; fatty acids and derivatives thereof, in particular, fatty acid monoglyceride polyglycol ethers; cross-linked polyacrylates such as Carbopol™, (polymers available from Goodrich); acrylates and copolymers thereof; polyvinylpyrrolidone and copolymers thereof; polyethylene 10 imines; salts such as sodium chloride and ammonium sulphate; sucrose esters; gellants; and mixtures thereof.

Of the clays, particularly preferred are synthetic hectorite (laponite) clay used in conjunction with an electrolyte salt 15 capable of causing the clay to thicken. Suitable electrolytes include alkali and alkaline earth salts such as halides, ammonium salts and sulphates.

The surfactant phase may also comprise a thickening agent, 20 i.e., a material which maintains the viscosity of this phase as the shear rate thereof is increased during use. Suitable materials include cross-linked polyacrylates such as Carbopol™ polymers available from Goodrich); natural gums including alginates, guar, xanthan and polysaccharide 25 derivatives including carboxy methyl cellulose and hydroxypropyl guar; propylene glycols and propylene glycol oleates; salts such as sodium chloride and ammonium sulfate; glycerol tallowates; and mixtures thereof.

- 12 -

It should be appreciated that, although there is a separate benefit agent stripe (i.e., the multiple emulsion stripe  $W_1-O-W_2$  defined by (B)), some benefit agent (e.g., hydrophobic agents such as silicone, petrolatum; or 5 hydrophilic benefit agents or water soluble agents such as polyglycerin) may also be added to the surfactant stripe. Other components which may be used in the surfactant stripe are soluble/salts (e.g., alkali metal chloride), for example, to control viscosity; small amounts of opacifiers 10 (preferably 0.2 to 2%), preservatives (0.2 to 2.0 wt. %), perfumes (0.5 to 2.0 wt.%) etc.

The separate benefit agent stripe is a multiple emulsion stripe which is similar to the stable emulsion described in 15 "Phase Stable Multiple Emulsion Compositions" to Naser et al., filed on same date as subject application and hereby incorporated by reference into the subject application.

The multiple emulsion composition comprises both a water in 20 oil portion and an external water emulsion surrounding the water in oil emulsion. More specifically, the multiple emulsion comprises:

- (1) 1 to 99% of a  $W_1/O$  emulsion which itself comprises
  - (a) about 1 to 99% internal aqueous phase containing water, 0.01 to 30% optional solute and 0 to 30% optional surfactant;
  - (b) 0.5 to 99% oil phase O surrounding said internal aqueous phase comprising a non-volatile silicone compound, a volatile

- 13 -

hydrocarbon compound, a non-volatile hydrocarbon compound or mixtures thereof;

(c) about 0.1 to 20% of a low HLB emulsifier; and

(d) topically effective amount of a water-soluble benefit agent in the internal water phase;

5 and

(2) 1 to 99% of an external aqueous phase W<sub>2</sub> which comprises:

- 10 (a) 0 to 30%, preferably 0.01 to 5% of high HLB nonionic surfactant emulsifier surfactant;
- (b) 0-60% of a second topically active compound;
- (c) optional solute; and
- (d) 0-20% optional cleansing surfactant or surfactants wherein, if anionic, the
- 15 surfactant is a non-amido surfactant.

The water phase of the W<sub>1</sub>/O part of emulsion will generally comprise droplets with an internal aqueous phase containing water, optional solute (preferably 0.01 to 30%) and a topically active compound (e.g., glycolic and/or lactic acid); and an oil phase surrounding said internal aqueous phase. Droplets may range from 0.1 to about 10 microns and may be enveloped by a membrane or film comprising oil and low HLB emulsifier.

25

The W<sub>1</sub>O emulsion is present in an amount 1-99%, preferably 2 to 90%, more preferably 5 to 90%, most preferably 10 to 70% by wt. of multiple emulsion W<sub>1</sub>-O-W<sub>2</sub>.

As noted, W<sub>1</sub> aqueous phase comprise 1 to 99% by wt. of the emulsion with small amounts of oil and low HLB emulsifier separating W<sub>1</sub> and W<sub>2</sub>. The water phase may comprise water, solute and water-soluble topically active compound (oil  
5 soluble active may also be in oil phase). It may also comprise additional active compounds and/or optional water soluble compounds providing desired aesthetic or functional effect (e.g., perfume).

- 10 The W<sub>1</sub> phase may comprise 1-99%, preferably 10-95% of W<sub>1</sub>/O emulsion, more preferably 25 to 95% of emulsion.

The topically active compound may enter the water or oil phase depending on solubility (e.g., water soluble compound  
15 in water phase and oil soluble in oil phase).

As used herein, the term "water soluble" means water soluble or water dispersible. A water soluble compound has a water solubility of at least 0.1 g (grams) per 100 ml  
20 (milliliters) of water and forms a true solution. A water soluble compound can be inherently water soluble or can be made water soluble by the addition of a solubilizing compound, such as a coupling agent, a co-surfactant, or a solvent. A water dispersible compound remains dispersed in  
25 water for at least the time period necessary to manufacture the primary W<sub>1</sub>/O emulsion, i.e., at least about one hour.

- 15 -

In addition, the topically active compound can be incorporated into the external aqueous phase W<sub>2</sub> to achieve enhanced efficacy.

- 5    The topically active compound therefore can be one of, or a combination of, a cosmetic compound, a medicinally active compound or any other compound that is useful upon topical application to the skin or hair. Such topically active compounds include, but are not limited to, hair conditioners  
10 (e.g., water soluble quaternary ammonium compounds, alkoxylated and nonalkoxyated fatty amines, dimethicone copolyols, cationic polymers etc.), skin conditioners (e.g., humectants such as glucose, glycerin, propylene glycol, amino acids, vitamins, amino functional silicones, etc.),  
15 hair and skin cleansers, hair fixatives, hair dyes, hair growth promoters, deodorants (e.g., organic and inorganic salts of aluminum, zirconium, zinc and mixtures thereof), skin care compounds, permanent wave compounds, hair relaxers, hair straighteners, antibacterial compounds,  
20 antifungal compounds, anti-inflammatory compounds, topical anesthetics, sunscreens and other cosmetic and medicinal topically effective compounds (e.g., antifungal, antibacterial).  
  
25   Sufficient water is present in the aqueous phase such that the aqueous phase comprises about 1% to about 95% by weight of the primary emulsion. Total water present in the W<sub>1</sub>-O-W<sub>2</sub> multiple emulsion composition is about 30% to about 99.9%, and typically about 40% to about 95%, by weight of the  
30 composition.

- 16 -

Among optional ingredients which may be included in the internal aqueous phase are included solutes. Among solutes which may be added are salts such as alkali metal chloride. Solute added to the internal aqueous phase may comprise 0 to 5 30% by wt., preferably 0.1-10% by wt.

It is also possible to add surfactant to the internal aqueous phase though this is added generally to the external W<sub>2</sub> phase, if it is to be added at all. The surfactant can 10 be any of the surfactants discussed in connection with the W<sub>2</sub> phase below.

The internal aqueous phase also can include optional ingredients traditionally included in topically applied 15 compositions. These optional ingredients include, but are not limited to, dyes, fragrances, preservatives, antioxidants, detackifying agents, and similar types of compounds. The internal phase may also include polyalkylene oxide components. These may be particularly useful for 20 incorporating certain hydrophilic benefit agents (e.g., salicylic acid) which may otherwise be difficult to add to the internal aqueous phase. The optional ingredients are included in the internal aqueous phase of the primary emulsion in an amount sufficient to perform their intended 25 function.

The oil phase may be volatile (except for silicone) or non-volatile.

- 17 -

A volatile oil may comprise a volatile hydrocarbon oil which evaporates during the process of drying skin or hair, and thereby releases the internal aqueous W<sub>1</sub> phase, which includes the first topically active compound to contact the 5 skin or hair. This oil preferably should contain no volatile silicone since volatile silicones tend to destabilize the multiple emulsion.

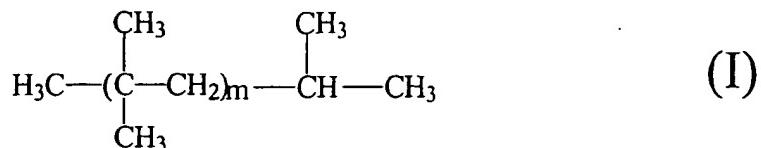
In one embodiment, the oil may comprise a combination of a 10 volatile oil (except for silicone) and a nonvolatile oil. In this embodiment, an oil can be designed to evaporate at a pre-selected temperature and provide a controlled release of the first topically active compound at the pre-selected temperature. Pre-selected temperatures are those 15 encountered during normal hair drying, provided by a hair dryer, or provided by a curling iron.

As previously stated, the oil also can include a water insoluble topically active compound in a sufficient amount 20 to impart a particular functional or aesthetic effect (e.g., emolliency), as long as the topically active compound does not adversely affect the W<sub>1</sub>-O-W<sub>2</sub> multiple emulsion composition (e.g., does not impart emulsion instability).

25 Volatile hydrocarbon compounds incorporated into the primary emulsion include, for example, isododecane and isohexadecane, i.e., PERMETHYL 99A, PERMETHYL 101A and PERMETHYL 102A, available from Presperse, Inc., South Plainfield, N.J. Other exemplary volatile hydrocarbon

- 18 -

compounds are depicted in general structural formula (I), wherein n ranges from 2 to 5.



5

Another exemplary volatile hydrocarbon compound is ISOPAR M (a C<sub>13</sub>-C<sub>14</sub> isoparaffin available from Exxon Chemical Co., Baytown, Tex). Preferably, the volatile hydrocarbon is less than 50% unsaturated.

10

As previously stated, the oil also can be a non-volatile oil. The non-volatile oil comprises a non-volatile silicone compound, a non-volatile hydrocarbon, or a mixture thereof. Preferably, the non-volatile oil comprises compounds which 15 contain less than 50% unsaturation. The non-volatile oil does not evaporate from the skin or hair. The first topically active compound therefore is released by rubbing the skin or hair to rupture the primary W/O emulsion. A non-volatile oil phase has a boiling point at atmospheric 20 pressure of greater than about 250°C.

Exemplary non-volatile silicone compounds include a polyalkyl siloxane, a polyaryl siloxane or a polyalkylaryl siloxane. Mix of these non-volatile silicone compounds also 25 are useful.

The non-volatile oil also can comprise a non-volatile hydrocarbon compound, such as mineral oil, petrolatum,

- 19 -

sunflower seed oil, canola oil or mixtures thereof. Other exemplary non-volatile hydrocarbon compounds that can be incorporated into the oil phase include, but are not limited to, a branched 1-decene oligomer, like 1-decene dimer or a 5 polydecene.

The oil also optionally can comprise (1) an oil, such as jojoba oil, wheat germ oil or purcellin oil; or (2) a water insoluble emollient, such as, for example, an ester having 10 at least about 10 carbon atoms, and preferably about 10 to about 32 carbon atoms.

Suitable esters include those comprising an aliphatic alcohol having about eight to about twenty carbon atoms and 15 an aliphatic or aromatic carboxylic acid including from two to about twelve carbon atoms, or conversely, an aliphatic alcohol having two to about twelve carbon atoms with an aliphatic or aromatic carboxylic acid including about eight to about twenty carbon atoms. The ester is either straight 20 chained or branched. Preferably, the ester has a molecular wt. of less than about 500. Suitable esters therefore include, for example, but are not limited to:

- (a) aliphatic monohydric alcohol esters (e.g.,  
25 isopropyl isostearate, cetyl acetate, cetyl stearate);  
myristyl propionate,  
isopropyl myristate,  
isopropyl palmitate,  
30 cetyl acetate,  
cetyl propionate,

- 20 -

cetyl stearate,

(b) aliphatic di- and tri-esters of polycarboxylic acids,  
(e.g., diisopropyl adipate);

5

(c) aliphatic polyhydric alcohol esters (e.g., propylene  
glycol dipelargonate); and

10 (d) aliphatic esters of aromatic acids, (e.g., C<sub>12</sub>-C<sub>15</sub>  
alcohol esters of benzoic acid).

The oil phase of the present invention also may include  
about 0.1% to about 30%, and preferably about 0.1% to about  
15%, on the weight of the oil of a low HLB emulsifier.

15

The low HLB emulsifier may comprise a silicon-free  
surfactant, or a blend of silicon-free surfactants, having  
an HLB value of about 10 or less (i.e., an HLB value of  
about 0.1 to about 10), an oil-soluble silicon-based  
20 surfactant, an oil-soluble polymeric surfactant, or mixtures  
thereof. Preferably, the silicon-free surfactant or  
surfactant blend has an HLB value of about 1 to about 7. To  
achieve the full advantage of the present invention, the  
silicon-free surfactant or surfactant blend has an HLB value  
25 of about 3 to about 6. The term "oil-soluble" as used  
herein means a compound having a solubility of at least 0.1  
g per 100 ml of oil phase to form a true solution.

30 The HLB value of a particular silicon-free surfactant can be  
found in *McCutcheon's Emulsifiers and Detergents, North  
American and International Editions*, MC Publishing , Glen

Rock, NJ (1993) (hereinafter McCutcheon's). Alternatively, the HLB value of a particular surfactant can be estimated by dividing the weight percent of oxyethylene in the surfactant by five (for surfactants including only ethoxy moieties).

- 5 In addition, the HLB value of a surfactant blend can be estimated by the following formula:

$$\text{HLB} = (\text{wt.\% A}) (\text{HLB}_A) + (\text{wt.\% B}) (\text{HLB}_B),$$

- 10 wherein wt. % A and wt. % B are the weight percent of surfactants A and B in the silicon-free surfactant blend, and  $\text{HLB}_A$  and  $\text{HLB}_B$  are the HLB values for surfactants A and B, respectively.

- 15 Low HLB surfactant can be a silicone-based surfactant or silicone-free surfactant.

- Exemplary classes of silicon-free nonionic surfactants include, but are not limited to, polyoxyethylene ethers of fatty (C<sub>6</sub>-C<sub>22</sub>) alcohols, polyoxyethylene/polyoxypropylene ethers of fatty (C<sub>6</sub>-C<sub>22</sub>) alcohols, ethoxylated alkylphenols, polyethylene glycol ethers of methyl glucose, polyethylene glycol ethers of sorbitol, and mixtures thereof.

- 25 Exemplary silicon-free nonionic surfactants are the ethoxylated alcohols having an HLB value of about 0.1 to about 10. An especially preferred ethoxylated alcohol is laureth-1, i.e., lauryl alcohol ethoxylated with an average of one mole of ethylene oxide. Other suitable ethoxylated alcohols include laureth-2, laureth-3 and laureth-4.

- 22 -

Numerous other nonionic surfactants having an HLB of about 0.1 to about 10 are listed in McCutcheon's at pages 229-236, incorporated herein by reference. Other exemplary silicon-free nonionic surfactants having an HLB value of about 0.1  
5 to about 10 include, but are not limited to, the ethoxylated nonylphenols, ethoxylated octylphenols, ethoxylated dodecylphenols, ethoxylated fatty (C<sub>6</sub>-C<sub>22</sub>) alcohols having four or fewer ethylene oxide moieties, oleth-2, steareth-3, steareth-2, ceteth-2, oleth-3, and mixtures thereof.

10

The emulsifier can also comprise a silicon-free surfactant blend having an HLB value of about 1 to about 10. The blend is a mixture of a sufficient amount of a surfactant having a low HLB value, i.e., about 0.1 to about 10, and a  
15 sufficient amount of a surfactant having a higher HLB value, i.e., about 1 to greater than about 10, such that the surfactant blend has an HLB value of about 1 to about 10. Exemplary, but non-limiting, nonionic surfactants having a high HLB value are listed in McCutcheon's at pages 236-246,  
20 incorporated herein by reference.

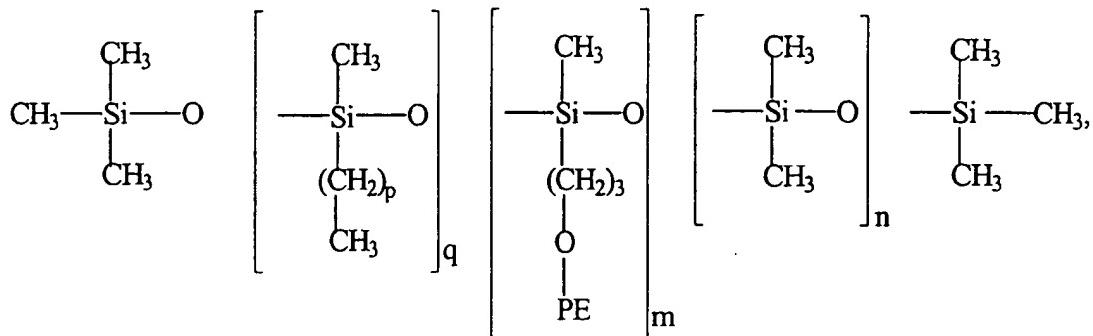
A preferred silicone-free surfactant is PEG-30 dipolyhydroxystearate.

25 An exemplary oil-soluble silicon-based surfactant is a dimethicone copolyol, which is a dimethylsiloxane polymer having polyoxyethylene and/or polyoxypropylene side chains, such as DOW CORNING 322C FORMULATION AID, available from Dow Corning Co., Midland, Mich. The dimethicone copolyol has  
30 about 15 or fewer ethylene oxide and/or propylene oxide monomer units, in total, in the side chains. Dimethicone

- 23 -

copolymers conventionally are used in conjunction with silicones because the silicon-containing surfactants are extremely soluble in a volatile or a non-volatile silicone compound, are extremely insoluble in water, and have a low 5 skin irritancy potential.

Another exemplary, but non-limiting, oil-soluble, silicon-based surfactant is an alkyl dimethicone copolyol, such as cetyl dimethicone copolyol available commercially as ABIL<sup>(R)</sup> 10 EM 90 from Goldschmidt Chemical Corporation, Hopewell, Va. The alkyl dimethicone copolyols have the structure:



15

wherein:

p is a numeral from 7 through 17;  
 q is a numeral from 1 through 100;  
 20 m is a numeral from 1 through 40;  
 n is a numeral from 0 through 200; and

- 24 -

PE is  $(C_2H_4O)_a(C_3H_6O)_b-H$  having a molecular weight of about 250 to about 2000, wherein a and b are selected such that the weight ratio of  $C_2H_4O/C_3H_6O$  is from 100/0 to 20/80.

5 The emulsifier may also be an oil-soluble polymeric surfactant. Polymeric surfactants capable of forming water-in-oil emulsions completely cover the surface of the water droplet, are firmly anchored at the oil-water interface, the external oil phase is a good solvent for the stabilizing  
10 portion of the polymeric surfactant, and the thickness of the polymer layer on the oil side of the interface is sufficient to ensure stability. These surfactants may include ethoxy, propoxy and/or similar alkylene oxide monomer units, e.g., butoxy.

15 The oil-soluble polymeric surfactants act as surfactants and are not physically or chemically cross-linked in solution. The oil-soluble polymeric surfactants therefore are differentiated from polymeric gelling agents such as  
20 polyacrylic acid or polymethacrylic acid.

Accordingly, exemplary oil-soluble polymeric surfactants include, but are not limited to, polyoxyethylene-polyoxypropylene block copolymers, and similar  
25 polyoxyalkylene block copolymers. The oil-soluble block copolymers typically have less than about 20% by weight of ethylene oxide. Specific non-limiting oil-soluble polymeric surfactants include Poloxamer 101, Poloxamer 105, PPG-2-Buteth-3, PPG-3-Butech-5, PPG-5-Butech-7, PPG-7-Butech-10,  
30 PPG-9-Buteth-12, PPG-12-Buteth-16, PPG-15-Buteth-20, PPG-20-

- 25 -

Buteth-30, PPG-24-Buteth-27, PPG-28-Buteth-35, and PEG-15 Butanediol. Other useful oil-soluble polymeric surfactants are polyamines, i.e., polyoxyethylene-polyoxypropylene block copolymers of ethylene diamine, having less than about 40%  
5 by weight ethylene oxide.

In accordance with an important feature of the present invention, the hydrophobic moiety of a silicon-free surfactant, silicon-containing surfactant or polymeric  
10 surfactant is sufficiently soluble in the oil phase such that a sufficient amount of the surfactant is present in the oil phase to stabilize the primary W<sub>1</sub>/O emulsion.

In one embodiment when the oil phase comprises a silicone  
15 compound, the surfactant phase comprises either a silicon-based surfactant, a silicon-free surfactant having a hydrophobic moiety preferably containing about ten to about fourteen carbon atoms, an oil-soluble polymeric surfactant, or a mixture thereof. If the hydrophobic moiety of the  
20 silicon-free surfactant is saturated and includes more than about 14 carbon atoms, the silicon-free surfactant is insoluble in the silicone phase and the primary W<sub>1</sub>/O emulsion is unstable. If the hydrophobic moiety includes less than about 10 carbon atoms, the primary W<sub>1</sub>/O emulsion  
25 has a tendency to coalesces i.e., the emulsion droplets fuse to form large droplets.

The amount of surfactant phase necessary to provide a primary emulsion of desired W<sub>1</sub>/O droplet diameter varies

- 26 -

with the amount of aqueous phase in the primary emulsion and is easily determined by those skilled in the art.

One particularly preferred emulsifier is cetyl dimethicone  
5 copolyol.

The external aqueous phase ( $W_2$ ) of the  $W_1-O-W_2$  multiple emulsion may comprise an isotropic mixture of surfactants phase (as opposed to lamellar phase of U.S. Patent No. 10 5,656,280), if surfactants are present. The phase may also comprise 0-60% by wt. of phase, preferably 10-50% of a second topically active compound, similar to or different from the first topically active compound and optional solute and optional (preferably 0.01 to 10%) of a stabilizing 15 natural gum polymer. If present, surfactant may be a non-amido anionic or all amphoteric or mixtures of the two. Surfactant may be absent, but will typically comprise 2-80% of aqueous phase.

20 This external phase may comprise a mixture of anionic and/or nonionic (or other) surfactants in combination with amphoteric surfactants wherein anionic/nonionic comprises 1 to 99% of the surfactant system and amphoteric comprises 1 to 99% of the surfactant system, i.e., one should not 25 ideally be present to the exclusion of the other.

In general, surfactant can be any of the surfactants described in connection with the surfactant stripe phase.

- 27 -

More specifically, if surfactant is entire or majority (>60%) anionic, it is preferred nonionic gum polymer be used.

5 Particularly soluble and dispersible in anionic surfactants and anionic rich surfactant systems are the nonionic seed polysaccharides. One such stabilizer is guar gum, which is structurally composed of a straight chain of D-mannose with a D-galactose side chain on approximately every other  
10 mannose unit. The usual ratio of mannose to galactose is approximately 2:1 and molecular weights are usually on the order of 100,000 to 1,000,000. Another useful stabilizer is locust bean gum, a galactomannan consisting of a main chain of D-mannose units with single galactose side chains on  
15 approximately every fourth unit. As with guar gum, approximate molecular weights are between 100,000 and 1,000,000.

In systems where the external surfactant W2 is comprised  
20 entirely of or contains a majority (50% or higher) of amphoteric surfactant, an anionic gum polymer is preferred. These can be seaweed polysaccharides, exudate polysaccharides, or microbial polysaccharides.

25 A preferred anionic seaweed polysaccharide is carrageenan, or "Irish moss," a complex mixture of sulfated polysaccharides. Carrageenan is a mixture of galactans that carry varying proportions of half-ester sulfate groups linked to one or more of the hydroxyl groups of the  
30 galactose units, which are joined by alternating  $\alpha$ -1,3 and

$\beta$ -1,4 glycosidic linkages. The molecular weight usually ranges from 100,000 to 1,000,000.

A preferred anionic exudate polysaccharide is gum  
5 tragacanth, a complex mixture of acidic polysaccharides containing galacturonic acid, galactose, fucose, xylose, and arabinose. Another preferred anionic exudate polysaccharide is gum karaya, a partially acetylated high molecular weight polysaccharide which contains L-rhamnose, D-galactose, D-  
10 galacturonic acid, and D-glucuronic acid residues.

Most preferred in amphoteric surfactants or amphoteric rich surfactant systems are microbial anionic hetero-polysaccharide gums, most preferably xanthan gum. The  
15 primary structure of these microbial gums contains two glucose units, two mannose units, and one glucuronic acid unit.

According to the invention, the benefit agent stripe (with  
20 multiple emulsion) and surfactant stripe are separate but combinedly dispensable from a packaging means and typically a single packaging means. Such a packaging means includes those systems which comprise two separate compartments. Ensuring that the surfactant and benefit agent are separate  
25 can be achieved in a variety of ways. Packaging of the composition such that the surfactant and benefit agent are presented in separate compartments or in separate domains within the packaging; including encapsulation of the benefit agent; and by processing of the composition by co-extrusion  
30 to produce a striped product in which individual stripes contain either the surface active agent or benefit agent.

Compositions of the invention may be formulated as products for washing the skin, for example, bath or shear gels, hand washing compositions or facial washing liquids; pre- and 5 post-shaving products; rinse-off, wipe-off and leave-on skin care products; products for washing the hair and for dental use.

The compositions of the invention will generally be pourable 10 liquids or semi-liquids, e.g., pastes and will have a viscosity in the range of 250 to 100,000 mPas measured at a shear rate  $10\text{s}^{-1}$  and 25°C in a Haake rotoviscometer RV20.

When the product is formulated as a shower gel the viscosity 15 will generally be in the range 200 to 15000 mPas, preferably 500 to 15,000 measured at a shear rate  $10\text{s}^{-1}$  and 25°C.

When the product is formulated as a facial wash product the viscosity will generally be in the range 3000 to 100,000 20 mPas measured at a shear rate  $10\text{s}^{-1}$  and 25°C.

Other typical components of such compositions include opacifiers, preferably 0.2 to 2.0 wt.%; preservatives, preferably 0.2 to 2.0 wt.% and perfumes, preferably 0.5 to 25 2.0 wt.%.

Except in the operating and comparative examples, or where otherwise explicitly indicated, all numbers in this description indicating amounts or ratios of materials or 30 conditions or reaction, physical properties of materials

- 30 -

and/or use are to be understood as modified by the word "about".

- Where used in the specification, the term "comprising" is
- 5 intended to include the presence of stated features, integers, steps, components, but not to preclude the presence or addition of one or more features, integers, steps, components or groups thereof.
- 10 The following examples are intended to further illustrate the invention and are not intended to limit the invention in any way.

All percentages used, unless indicated otherwise, are

15 intended to be percentages by weight.

Glycolic Acid and Sodium Glycolate Deposition Protocol

Procedure for Analyzing the Deposition of Glycolic Acid and

20 Sodium Glycolate from Cleansers by Gas Chromatography (GC)

Test Procedure (Pigskin):

Fresh Piglet skin (4-5 weeks old, male, white, no nipples

25 and shaved) is used for the deposition studies. The skin is cut into ~12 cm x 5 cm pieces which are stretched and pinned onto a support. The surface of the skin is wetted with 250 ml water and 1 gram of cleanser is applied to the wet surface (0.5 gram benefit stripe and 0.5 gram of surfactant

30 phase). The skin is then washed by hand for 30-60 seconds,

- 31 -

rinsed three times with 250 ml aliquots of water and patted dry with Kimwipe paper towel.

A glass/plastic cylinder 3.3 cm in diameter is then placed  
5 in close contact with the skin surface and the skin is extracted three times with 3 ml aliquots of organic solvent (ethanol, acetone or tetrahydrofuran) followed by another three times with 3 ml aliquots of distilled water. The organic solvent extracts are combined. The same is done for  
10 the water extracts. (The water and/or organic solvent extractions can be repeated several times depending on the concentration of sodium glycolate or glycolic acid on the pigskin). The organic extract will contain the glycolic acid and the water will contain the sodium glycolate.

15 Three sites are extracted on each of three pieces of skin for each formulation. The concentration of the sodium glycolate and the glycolic acid on the skin (in  $\mu\text{g}/\text{cm}^2$ ) is determined by comparison with standard solutions on the Gas  
20 Chromatography and/or Liquid Chromatography using analytic procedure below.

Analytical Procedure

25 Apparatus 1:

A suitable Gas Chromatography (HP 5890 series II perhaps)

Sil-prep from Alltech (part #18013)

30

Gas Chromatography Setting:

- 32 -

Detector: Flame Ionization Detector  
Column: HP-1 100%-Polydimethylsiloxane,  
5 Column length 25.9 m,  
Inside diameter 0.2 mm, Film  
thickness 0.33 mm  
Injector Volume 4.0  $\mu$ L-5.0 $\mu$ L  
Gases: Helium (carrier), Hydrogen and Air  
(Flame)

10 Flow rate:  
Total flow rate:  
Detector temp.: 250°C  
Injector temp.: 250°C  
Oven temperature: Set point 50°C, limit 300°C  
15 Run time: 14.6 minutes

Oven Temperature Profile:

Initial temp.: 50°C  
20 Initial time: 0.0  
Rate: 25C/minute  
Final temp.: 290°C  
Final time: 14.6 minutes

25 Standards:

Solutions at 4000, 2000, 1000, 500, 250, 125, 62.5, 31.3,  
15.6, 7.8  $\mu$ g/ml are prepared. Glycolic acid in organic  
solvent. Sodium glycolate in water.

30

Apparatus 2:

- 33 -

A suitable Liquid Chromatography (HP)

Liquid Chromatography Settings:

5

Detector: UV Detector  
Column: Phenomenex LUNA 5u C18 (2)  
Injector volume: 5.0 uL-50.0 uL  
Mobil phase: 97% 20nM Potassium Phosphate @ pH  
10 2.3:3% Methanol  
Flow rate: 0.7 ml/minute  
Temperature: 25°C  
Run time: 30 minutes

15 Standards:

Solutions at 4000, 2000, 1000, 500, 250, 125, 62.5, 31.3,  
15.6, 7.8 µg/ml are prepared.

20 EXAMPLES

Example 1 - Preparation of Water-Oil-Water Benefit Stripe

25 Preparation of W<sub>1</sub>/O Phase of W<sub>1</sub>-O-W<sub>2</sub> Multiple Emulsion  
(Glycolic Acid Premix) Used as Benefit Stripe

Ingredients

- 34 -

- Glycolic Acid Solution (44% active @ pH 4)  
(benefit agent in internal aqueous phase);
- Petrolatum (oil phase)
- Abil EM90 Goldschmidt (cetyl dimethicone copolyol  
5 low HLB emulsifier);

### Preparation

In a beaker, petrolatum (oil phase) and EM90 (emulsifier)  
10 were added. Contents were heated to ~65-70°C (allowing  
petrolatum to melt). Contents were stirred at approximately  
800-900 rpms. Whilst stirring, glycolic acid solution  
(premix) was added dropwise. After addition of glycolic  
acid, the mixture was homogenized at 10,000 rpm for 10-15  
15 minutes and mixture was cooled overnight.

### Preparation of W<sub>1</sub>-O-W<sub>2</sub> Benefit Stripe

#### Ingredients

- 20
- Water;
  - Glycerin (optional);
  - Carbopol 980 (Any Carbomer may be used);
  - Cationic Polymer (e.g., Jaguar)
  - 25 - Biopolymers (Agar, Carragenaan);
  - Formaldehyde/Glydant II
  - Brij 58; (high HLB nonionic used in W<sub>2</sub> phase);
  - Borage Seed Oil (for oil phase);
  - 2M NaOH
  - 30 - Water-in-Oil (Glycolic Acid) Premix (Prepared as  
noted above)

Preparation

5      Beaker A: In a Beaker A, water-in-oil premix (prepared as noted above), Brij 58 and borage seed oil were added.

Contents were heated to ~65-70°C. Ingredients were stirred well.

10     Beaker B: In Beaker B, water, glycerin, formaldehyde and carbopol 980/cationic polymer/bipolymer were added. Contents were stirred at ~700 rpm until Carbopol 980 was fully dispersed in aqueous solution. The mixture was heated to ~65-70°C.

15     Beaker A was removed from heat and Beaker B was added to Beaker A. Contents were mixed till homogenous, 2M NaOH was added and mixed for 30-40 minutes @ 500 rpms. The contents were cooled overnight.

20     Example 2

Preparation of Benefit Agent PremixIngredients

- 25
- Glycolic Acid Solution (44% active @ pH 4);
  - Water;
  - NaOH @ 50%;
  - Glycolic Acid @ 99.0%;
  - 30      -      Carbomer

Preparation

Water was added to a beaker. Glycolic acid was added and stirred until glycolic acid had dissolved. Glycolic acid  
5 solution was cooled to 0°C. NaOH was slowly added (drop-wise). After addition of NaOH, ingredients were cooled to room temperature. Carbomer was added to solution slowly, stirring rapidly until Carbomer was fully dispersed. The solution became viscous.

10

Example 3

Following is an example of a stripe which was prepared to be used as surfactant stripe phase ((A)):

15

Ingredient	% By Wt.
Acyl Isethionate	1 - 15%
Betaine	5 - 20%
Nonionic	0 - 8%
NaOH	0 - 2%
Lytron 621 (Opacifier)	0 - 2%
Antil 141 (Thickener)	0 - 3%
Preservative	0 - 2%
Water	To balance

Example 4

The following W-O-W<sub>2</sub> Petrolatum Stripe and Base (Central  
20 Base, Stripe A) were prepared:

Stripe A - Base

- 37 -

Ingredient	% Active
Cocamidopropyl betaine	15.0
Sodium cocoyl isethionate	10.0
Sodium laureth sulfate	3.0
Cationic polymer	1.5
Glycerin	0-15
Coco monoethanolamide	0-10
Thickener	0.01-5
Fragrance, minors	0-1.0
Water	To balance

Stripe B - Benefit Agent Stripe

Ingredient	% Active
Petrolatum	36.6
Glycerin	10.0
Cetyl dimethicone copolyol	1-5.0
Borage seed oil	1.0
Hydroxyacetic acid (glycolic acid)	20.0*
Ceteth-20 (NI Surfactant)	0.1
Carbopol	1.0
Sodium hydroxide	5-6
DMDM Hydantoin	0.2
Water	To balance

- 5 \*In general in both base and stripe phases, levels come out as half of the amount when dispersed. Thus, 20% glycolic in benefit agent stripe is dispersed as 10% glycolic when out of the container.

10 Comparative A (Oil-In-Water Benefit Stripe)

The following dual stripe composition was also prepared but, in contrast to Example 4, the benefit gel stripe was not W-O-W<sub>2</sub> stripe with 10% glycolic acid but an oil-in-water stripe  
15 with 10% glycolic.

- 38 -

Stripe A - Base

Ingredient	% Active
Cocamidopropyl betaine	10.0
Sodium cocoyl isethionate	3.0
Sodium laureth sulfate	2.0
Thickener	0.0-5
Fragrance, minors	0-1.0
Water	To balance

5

Stripe B - Benefit Agent (Oil-In Water Emulsion)

Ingredient	% Active
Petrolatum	36.0
Borage seed oil	1.0
Hydroxyacetic acid (glycolic acid)	20.0
Ceteth 20 (NI Surfactant)	0.1
Carbopol	1.0
Sodium hydroxide	5-6
DMMD Hydantoin	0.2
Water	To balance

Benefit Agent in Comparative contains no glycerin.

10 Comparative B (Single Cleanser)

The following single cleanser system was prepared and was used in Example 5 set forth below:

Ingredient	% Active
Cocamidopropyl betaine	10.0
Sodium acyl isethionate	3.0
Sodium laureth sulfate	2.0
Borage seed oil	1.0
Hydroxyacetic acid	10.0
Sodium hydroxide	5-6
DMDM Hydantoin	0.36
Water	To balance

Example 5

5 Using the compositions of Example 4 and of the Comparative A (dual cleanser where glycolic acid is directly in water phase), these were compared for deposition results using protocol set forth previously and results are set forth below:

10

O/W Composition32 $\mu\text{g}/\text{cm}^2$ W/O/W Composition200  $\mu\text{g}/\text{cm}^2$ 

15 These paragraphs clearly show that a much greater amount of glycolic acid was deposited from multiple W/O/W emulsion system than from single O/W emulsion system.

Example 6

20

The applicants also compared deposition from W/O/W emulsion system from dual stripe cleanser system compared to deposition from a single stripe cleanser as set forth in Comparative B. These results are as follows:

- 40 -

<u>Single Stripe</u>	<u>Dual Stripe with W/O/W</u>
	<u>Multiple Emulsion in Benefit</u>
	<u>Stripe</u>

5

0.5  $\mu\text{g}/\text{cm}^2$

200  $\mu\text{g}/\text{cm}^2$

These paragraphs clearly show that a much greater amount of glycolic acid was deposited from multiple W/O/W emulsion system than from single system.

10

#### Example 7

The following multiple emulsion was also prepared using a  
15 surfactant stripe (Stripe A) and benefit agent stripe  
(Stripe B) as noted below:

#### Stripe A - Surfactant Base

20	<u>Ingredient</u>	<u>% Active in Stripe</u>
	Cocamidopropyl Betaine	15.6
	Sodium Laureth Sulfate	10.4
	Deposition Polymer (Cationic polymer)	0.9
	Water	73.1

25

- 41 -

Stripe B - Benefit Agent W<sub>1</sub>/OW<sub>2</sub> Stripe

	<u>Ingredient</u>	<u>% Active in Stripe</u>
	Mineral Oil (Light)	14.4
5	Sodium Lactate	7
	Sodium Chloride	4.2
	Cetyl Dimethicone Copolyol	3.6
	Cocamidopropyl Betaine	2.4
	Sodium Laureth Sulfate	1.6
10	Water	66.8

This emulsion is similar to that of Example 4. The surfactant stripe is essentially same (betaine/isethionate). The benefit agent stripe, however, uses sodium lactate as  
15 benefit agent and mineral oil as the oil.

This example shows that many different benefit agents can be used as part of the multiple emulsion.

CLAIMS

1. An aqueous liquid cleansing and moisturizing composition comprising:

5

- (A) 10 to 99.9% by wt. of a surfactant stripe comprising 1 to 75% by wt. of a surface active agent selected from anionic, nonionic, zwitterionic and cationic surface active agents, soap and mixtures thereof; and
- (B) 0.1 to 90% by wt. of a benefit agent stripe comprising a multiple emulsion where said multiple emulsion comprises:
  - (1) 1 to 99% of said multiple emulsion of a water-in-oil emulsion comprising:
    - (a) 1 to 99% of an internal aqueous phase containing water, optional solute and optional surfactant;
    - (b) 0.5 to 99% of an oil phase surrounding an internal aqueous phase comprising a non-volatile silicone compound, a volatile hydrocarbon compound, a non-volatile hydrocarbon or a mixture thereof;
    - (c) 0.1 to 20% of a surfactant emulsifier having HLB of below 10; and
    - (d) a topically effective amount of a water-soluble benefit agent in the internal aqueous phase; and
  - (2) 1 to 99% of an external aqueous emulsion comprising 0 to 30% nonionic surfactant with

10

15

20

25

30

- 43 -

HLB of 10-30 surfactant, 0-60% topically active compound, optional solute, and 0-20% cleansing surfactant or surfactant.

5 2. A composition according to claim 1, wherein the surfactant of (A) comprises 5 to 70% of the stripe.

3. A composition according to claim 1 or claim 2, wherein benefit agent can be added to surfactant Stripe (A).

10 4. A composition according to any of the preceding claims, wherein the water-in-oil emulsion is 2 to 90% of the W<sub>1</sub>-O-W<sub>2</sub> emulsion.

15 5. A composition according to any of the preceding claims, wherein the water-in-oil emulsion comprises 0.1 to 10% by wt. solute.

20 6. A composition according to any of the preceding claims, wherein the oil phase of the water-in-oil emulsion encapsulates internal aqueous phase to form droplets of about 5 to about 1000  $\mu$ .

25 7. A composition according to any of the preceding claims wherein emulsifier having HLB below 10 comprises 0.1 to 15% of the oil phase.

8. A composition according to claim 7, wherein the emulsifier is a dimethicone copolyol.

- 44 -

9. A composition according to any of the preceding claims, wherein topically effective compounds in the water-in-oil emulsion comprise glycolic acid.

5 10. A composition according to any of the preceding claims, wherein the external aqueous phase comprises 20-95% multiple emulsion.

10 11. A composition according to any of the preceding claims, wherein if cleansing surfactant (B)(2) is present, it is non-amido surfactant.

15 12. A composition according to any of the preceding claims, wherein the surfactant system of (B)(2) is greater than 60% anionic surfactant and nonionic gum polymer is present.

13. A composition according to any of claims 1-12, wherein the surfactant of (B)(2) is greater than 50% amphoteric surfactant and anionic gum polymer is present.

# INTERNATIONAL SEARCH REPORT

Int'l. Jonal Application No  
PCT/EP 00/09762

**A. CLASSIFICATION OF SUBJECT MATTER**  
IPC 7 A61K7/00

According to International Patent Classification (IPC) or to both national classification and IPC

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

CHEM ABS Data, EPO-Internal

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 97 17938 A (UNILEVER PLC, UNILEVER N.V.) 22 May 1997 (1997-05-22) claim 1 ----	1
A	WO 96 37420 A (UNILEVER PLC, UNILEVER N.V.) 28 November 1996 (1996-11-28) claim 1 ----	1
A	WO 96 02230 A (UNILEVER PLC, UNILEVER N.V.) 1 February 1996 (1996-02-01) claim 1 ----	1
A	US 4 254 105 A (H. FUKUDA ET AL.) 3 March 1981 (1981-03-03) claim 1; example 10 -----	1

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

° Special categories of cited documents :

- °A° document defining the general state of the art which is not considered to be of particular relevance
- °E° earlier document but published on or after the international filing date
- °L° document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- °O° document referring to an oral disclosure, use, exhibition or other means
- °P° document published prior to the international filing date but later than the priority date claimed

°T° later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

°X° document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

°Y° document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

°&° document member of the same patent family

Date of the actual completion of the international search

13 February 2001

Date of mailing of the international search report

23/02/2001

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  
Fax: (+31-70) 340-3016

Authorized officer

Glikman, J-F

# INTERNATIONAL SEARCH REPORT

Information on patent family members

Int. Application No

PCT/EP 00/09762

Patent document cited in search report	Publication date	Patent family member(s)		Publication date
WO 9717938	A 22-05-1997	AU 703204	B	18-03-1999
		AU 7213896	A	05-06-1997
		BR 9611445	A	23-03-1999
		EP 0861063	A	02-09-1998
WO 9637420	A 28-11-1996	AU 5145896	A	11-12-1996
		AU 718944	B	04-05-2000
		AU 5820596	A	11-12-1996
		BR 9608796	A	17-02-1999
		BR 9608807	A	17-02-1999
		CA 2221559	A	28-11-1996
		CA 2221575	A	28-11-1996
		CN 1191478	A	26-08-1998
		CN 1191515	A	26-08-1998
		WO 9637179	A	28-11-1996
		EP 0828473	A	18-03-1998
		EP 0828670	A	18-03-1998
		JP 11505811	T	25-05-1999
		JP 11508154	T	21-07-1999
		US 5914116	A	22-06-1999
WO 9602230	A 01-02-1996	AU 2985895	A	16-02-1996
		BR 9508321	A	06-01-1998
		CA 2190738	A	01-02-1996
		EP 0755243	A	29-01-1997
		JP 2977908	B	15-11-1999
		JP 9512826	T	22-12-1997
		KR 249451	B	01-04-2000
		US 5612307	A	18-03-1997
		ZA 9505972	A	20-01-1997
US 4254105	A 03-03-1981	JP 53031578	A	24-03-1978
		JP 58006530	B	04-02-1983
		JP 1117081	C	15-10-1982
		JP 52046382	A	13-04-1977
		JP 55023087	B	20-06-1980
		FR 2326914	A	06-05-1977
		GB 1541463	A	28-02-1979